

## $\alpha$ -Onocerin chloroform hemisolvate

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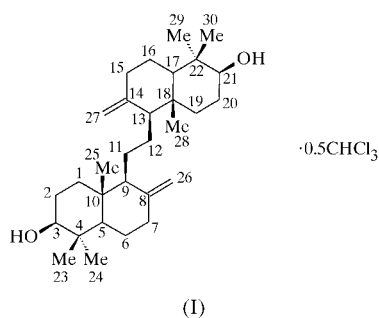
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The triterpenoid natural product  $\alpha$ -onocerin [8,14-secogam-macera-8(26),14(27)-diene-3,21-diol], determined here as the chloroform hemisolvate,  $C_{30}H_{50}O_2 \cdot 0.5CHCl_3$ , consists of two independent symmetric *trans*-decalin  $C_{15}$  building blocks. Hydrogen bonds between the hydroxyl groups form an infinite two-dimensional network perpendicular to the *c* axis.

### Comment

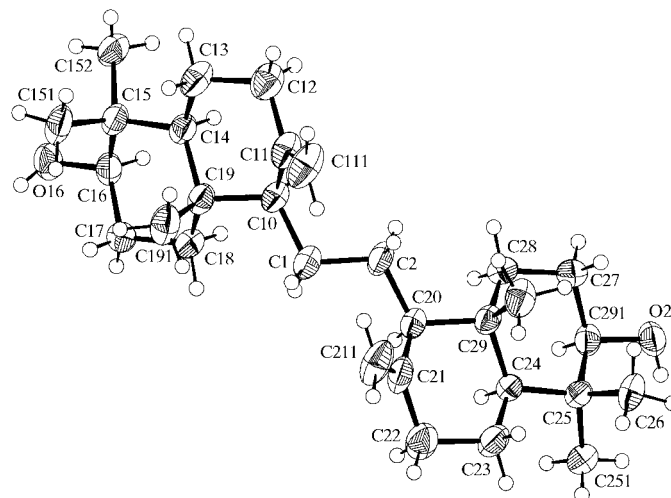
The roots of *Ononis spinosa* L. (Fabaceae) are used medicinally as a diuretic drug. Phytochemical identification, diagnosis of drug adulterations and the standardization of extracts are performed using pure  $\alpha$ -onocerin as a reference compound. While isolated in 1855, its constitution was assigned using chemical methods only in 1955 (Barton & Overton, 1955, and references therein). However, only the structure of the related 8(26),14(27)-diketones have been studied by X-ray analysis so far (Tsuda *et al.*, 1983). We report here the first X-ray structure of a 3,21-onoceradiene-3,21-diol, (I). This is in continuation of the comprehensive spectral evaluation of  $\alpha$ -onocerin focusing on NMR and MS spectral methods which was published recently (Pauli, 2000) and which supported the presence of two *trans*-decalin  $C_{15}$  building blocks symmetrically linked by a bimethylene bridge.



The structural investigation of (I) confirms the assumed geometry. The framework consists mainly of  $Csp^3-Csp^3$  bonds [1.498 (6)–1.576 (5) Å], with only C11–C111 and

C21–C211 [1.317 (7) and 1.329 (7) Å] being clearly  $Csp^2-Csp^2$  bonds. Also, both bonds to the hydroxyl groups are in the expected range for a single C–O bond [1.445 (5) and 1.441 (5) Å]. Furthermore, the geometries of both *trans*-decalin  $C_{15}$  building blocks are quite similar and comparable with the same skeleton found in gumozine (Nasirov *et al.*, 1977) and 3- $\alpha$ -hydroxymanool hydrate (Kagawa *et al.*, 1993).

Hydrogen bonds between the hydroxyl groups form an infinite two-dimensional network perpendicular to the *c* axis (Table 1). Besides these, there are no further contacts closer than van der Waals distances.



**Figure 1**  
*DIAMOND* (50% probability) plot (Brandenburg, 1996) of the title compound with the atomic numbering scheme.

The recently published NMR spectroscopic evidence for a symmetric constitution of  $\alpha$ -onocerin is now definitively corroborated by the crystallographic data. Therefore, it becomes clear that the  $C_{30}$  skeleton consists of two halves that are stereochemically identical. Compared with typical pentacyclic triterpene skeletons, such as oleanolic acid, however, this must be rated a very unusual feature with respect to the designated triterpenoid origin of  $\alpha$ -onocerin which has not been supported by biogenetic studies so far. It must be mentioned that no diastereomeric analogues, *e.g.* isomers with inversion of one or more stereocentres, could be detected by NMR (Pauli, 2000). The biogenetic assignment may be challenged in three points: the necessity for fully stereospecific alterations such as (i) the hydrogenation of the 12,13 double bond and (ii) the hydroxylation of the C-21 position, and most importantly, (iii) an inversion of the decalin ring fusion being [*D/E*]*cis* in oleanolic acid. Therefore, the classification of  $\alpha$ -onocerin as a dimeric sesquiterpene is equally justified at this point and should not be omitted. Concerning the numbering of the  $\alpha$ -onocerin framework, which in the literature is again based on the assumption that it represents a triterpene, at this point, revision should be avoided unless the biosynthesis has been investigated.

## Experimental

The title compound was obtained from the dried roots of *Ononis spinosa* (sample No. 9200280, PhytoLab, Vestenbergsgrueuth) upon Soxhlet extraction with petrol ether, cleaning-up with active charcoal, *in vacuo* precipitation, and repeated crystallization from petrol ether, CH<sub>2</sub>Cl<sub>2</sub>/EtOAc (1:1), and CHCl<sub>3</sub> in a final yield of 0.05%. Colourless crystals were obtained from CHCl<sub>3</sub>. Calculated mass for C<sub>30</sub>H<sub>50</sub>O<sub>2</sub>: 442.7. Mass spectrum (EI; *m/z*): 442 (*M*<sup>+</sup>), 427 ([*M*-CH<sub>3</sub>]<sup>+</sup>), 409 ([*M*-CH<sub>3</sub>-H<sub>2</sub>O]<sup>+</sup>), 391 ([*M*-CH<sub>3</sub>-2H<sub>2</sub>O]<sup>+</sup>), 381 ([*M*-CH<sub>3</sub>-CO-H<sub>2</sub>O]<sup>+</sup>). <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ, 1.572 (*qddd*, 1[19]A = eq), 0.952 (*ddd*, 1[19]B = ax), 1.552 (*dddd*, 2[20]A = eq), 1.441 (*dddd*, 2[20]B = ax), 3.090 (*dd*, 3[21]α), 0.946 (*dd*, 5[17]ax), 1.604 (*[d]dddd*, 6[16]A = eq), 1.239 (*dddd*, 6[16]B = ax), 2.268 (*ddd*, 7[15]A = eq), 1.839 (*dddddt*, 7[15]B = ax), 1.369 (*m*, 9[13]), 1.369 (*m[dddd]*, 11[12]A), 1.055 (*m[dddd]*, 11[12]B), 0.845 (*s*, 23[30]), 0.616 (*s*, 24[29]), 0.505 (*br s/d*, 25[28]), 4.695 (*dd*, 26[27]A), 4.442 (*ddd*, 26[27]B). <sup>13</sup>C NMR (CDCl<sub>3</sub>): 36.84 (1[19]), 27.27 (2[20]), 78.43 (3[21]), 38.84 (4[22]), 54.50 (5[17]), 23.81 (6[16]), 38.03 (7[15]), 148.28 (8[14]), 57.28 (9[13]), 39.00 (10[18]), 22.30 (11[12]), 27.91 (23[30]), 15.09 (24[29]), 14.21 (25[28]), 106.35 (26[27]).

### Crystal data

C <sub>30</sub> H <sub>50</sub> O <sub>2</sub> ·0.5CHCl <sub>3</sub>	Cu Kα radiation
<i>M<sub>r</sub></i> = 502.38	Cell parameters from 25 reflections
Orthorhombic, <i>P</i> 2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	<i>θ</i> = 22.46–46.14°
<i>a</i> = 7.470 (1) Å	<i>μ</i> = 1.716 mm <sup>-1</sup>
<i>b</i> = 14.681 (2) Å	<i>T</i> = 223 (2) K
<i>c</i> = 27.079 (4) Å	Block, colourless
<i>V</i> = 2969.7 (7) Å <sup>3</sup>	0.30 × 0.20 × 0.15 mm
<i>Z</i> = 4	
<i>D<sub>x</sub></i> = 1.124 Mg m <sup>-3</sup>	

### Data collection

Enraf–Nonius CAD-4 diffractometer	<i>θ</i> <sub>max</sub> = 74.29°
<i>ω/2θ</i> scans	<i>h</i> = 0 → 9
Absorption correction: empirical via <i>ψ</i> scan (Fair, 1990)	<i>k</i> = 0 → 18
<i>T</i> <sub>min</sub> = 0.627, <i>T</i> <sub>max</sub> = 0.783	<i>l</i> = 0 → 33
3450 measured reflections	3 standard reflections
3450 independent reflections	every 250 reflections
2318 reflections with <i>I</i> > 2σ( <i>I</i> )	frequency: 120 min
	intensity decay: 6.7%

**Table 1**

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O16—H16A...O26 <sup>i</sup>	0.83	2.04	2.822 (5)	157
O26—H26A...O16 <sup>ii</sup>	0.83	2.17	2.817 (5)	135

Symmetry codes: (i) *x*, *y* - 1, *z*; (ii)  $\frac{1}{2} + x$ ,  $\frac{3}{2} - y$ ,  $-z$ .

### Refinement

Refinement on <i>F</i> <sup>2</sup>	$w = 1/[\sigma^2(F_o^2) + (0.1687P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.076$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.228$	$(\Delta/\sigma)_{\max} = 0.001$
<i>S</i> = 0.991	$\Delta\rho_{\max} = 0.70 \text{ e \AA}^{-3}$
3450 reflections	$\Delta\rho_{\min} = -0.34 \text{ e \AA}^{-3}$
313 parameters	Absolute structure: Flack (1983)
H-atom parameters constrained	Flack parameter = -0.05 (19)

The asymmetric unit contains half a disordered molecule of chloroform as a residual solvent. This disorder decreases the quality of the whole investigation. The Flack parameter of -0.05 (19) favours the reported absolute configuration, but due to the disorder problem and the fact that a large fraction of Friedel-related reflections have not been measured, this detail should not be overrated. The occupancy of the disordered chloroform was first refined and then fixed with a site-occupancy factor of 0.5. Refinement with anisotropic displacement parameters and/or restraints did not improve the model; therefore, these four atoms are only refined isotropically.

Data collection: *EXPRESS* (Nonius, 1994); cell refinement: *EXPRESS*; data reduction: *MolEN* (Fair, 1990); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 1996); software used to prepare material for publication: *SHELXL97*.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1405). Services for accessing these data are described at the back of the journal.

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